

Strong Correlation Effects in Fullerene Molecules and Solids

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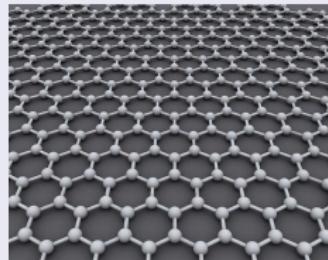
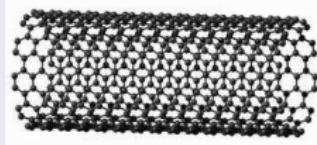
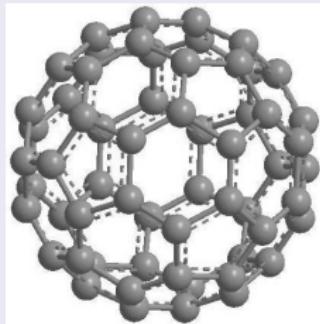
Outline:

- Introduction to Fullerene molecules and solids.
- Effective tight-binding Hubbard model for the molecules.
- C_{20} molecule and solid: pair-binding energies; magnetic properties; density of states, etc.
- C_{60} molecule and solid: Single-particle excitation spectrum for solid; molecular orientations; comparison with ARPES.
- Some recent experimental developments.
- Summary.

Introduction

Carbon based nano materials

- 0D: cage structures: C_{60} , C_{20} , etc. (Nobel Prize in 1996)
- 1D: rod structures: nanotube.
- 2D: planar sheet structure: graphene (Nobel prize 2010)



Introduction

Intensive researches in the past

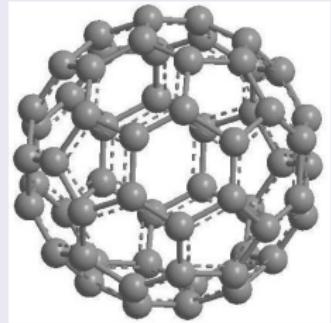
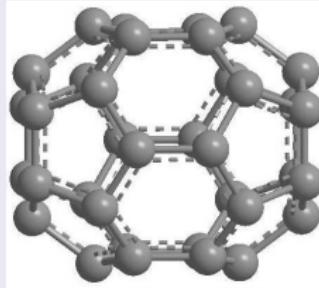
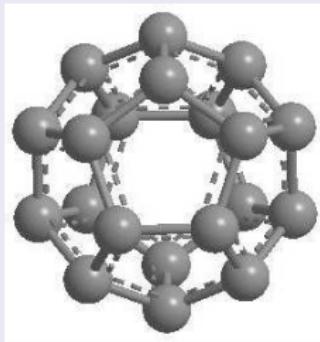
- 21,000 derivatives of fullerene found by the end of 2002;
- 23,000 papers published by the end of 2002;
- Amazing properties: superconducting, ferromagnetic, antiferromagnetic, spin density wave, ...
- Crystal structures seems relatively simpler than cuprates;
- However, after 20 years' research superconducting mechanism still unclear.

Introduction.

What are Fullerene molecules?

Carbon cage molecules with exactly 12 pentagons and varying number of hexagons. A consequence of Euler's polyhedral formula: $V - E + F = 2$.
Three examples:

- C_{20} : $V = 20$, $E = 30$, $F = 12$;
- C_{36} : $V = 36$, $E = 54$, $F = 20$;
- C_{60} : $V = 60$, $E = 90$, $F = 32$.



Introduction.

Synthesis and properties.

- C₆₀ discovered by Kroto, Curl, and Smalley [Nature, 1985] (Nobel Prize in 1996).
- K₃C₆₀ molecular superconductor (T_c=18K) found by Hebard et al [Nature, 1991].
- "Pentagon separation rule", Kroto [Nobel lecture, 1997].
- Solid C₃₆ first produced by Zettl group [Nature, 1998].
- Gas phase C₂₀ first produced by Prinzbach et al [Nature, 2000].
- Solid C₂₀ synthesized by Wang et al [Phys. Lett. A , 2001] and Iqbal et al [Eur. Phys. J. B, 2003].

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Effective Hamiltonian Approach

Importance of Carbon radial p_r orbitals

Fullerene molecules electronic properties are mainly determined by the electrons from Carbon atom radial p_r orbitals. Eg. for C₆₀ molecule, 94% of bonds around Fermi energy are made of p_r orbitals as shown by Satpathy [Chem. Phys. Lett., 1985].

→ Conjugated π bonds: electron delocalization

→ **One-band tight-binding** Hamiltonian description can be successful.

Effective Hamiltonian: Hubbard Model

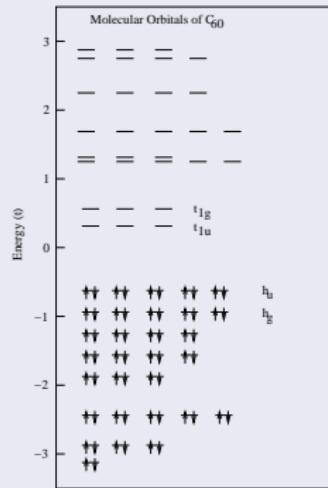
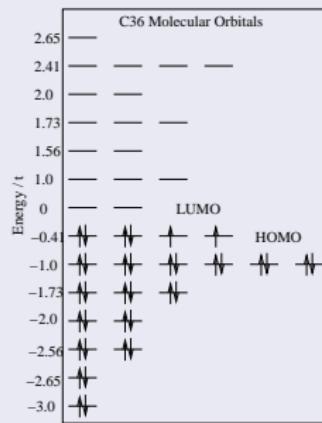
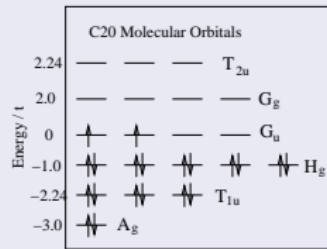
$$H = - \sum_{\langle ij \rangle \sigma} t_{ij} (c_{i\sigma}^\dagger c_{j\sigma} + h.c.) + U \sum_i n_{i\uparrow} n_{i\downarrow}, \quad (1)$$

- What are U/t values?
- Solve by quantum Monte Carlo (QMC) and exact diagonalization (ED).

Effective Hamiltonian Approach

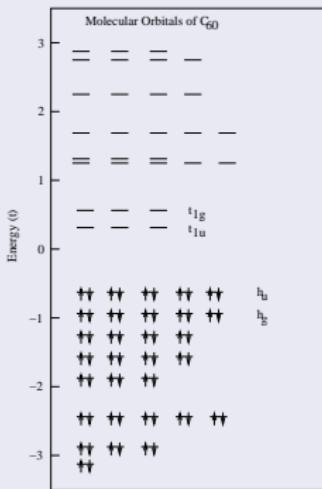
Huckel energy level diagrams

For $U = 0$ (non-interacting), we can easily diagonalize Hamiltonian:



Effective Hamiltonian Approach

Hückel energy level diagrams: electron dopings



- C_{60}^- : quasi 1D lattice, phase transition at 50K from 1D conductor to spin density wave ground state; Chauvet et al [Phys. Rev. Lett. 1994], Stephens et al [Nature 1994].
- C_{60}^{2-} : no; not yet realized.
- C_{60}^{3-} : yes; discuss this later.
- C_{60}^{4-} : insulator; Fleming et al [Nature 1991]
- C_{60}^{5-} : metal monolayer; Crommie group [Science 2005, Phys. Rev. Lett. 2007]
- C_{60}^{6-} : insulator; Zhou et al [Nature 1991], Stephens et al [Phys. Rev. B 1992]

Effective Hamiltonian Approach

Parameters Determination

- Hopping t_{ij} can be determined by fitting Huckel energy levels around Fermi energy to those from DFT calculations.
- U can be determined by comparison to some experiments:

Electron affinity energy

Wang et al [J. Chem. Phys., 1999] for C₆₀

Prinzbach et al [Nature, 2000] for C₂₀.

$$\text{AE} = E(C_N) - E(C_N^-). \quad (2)$$

X-ray scattering

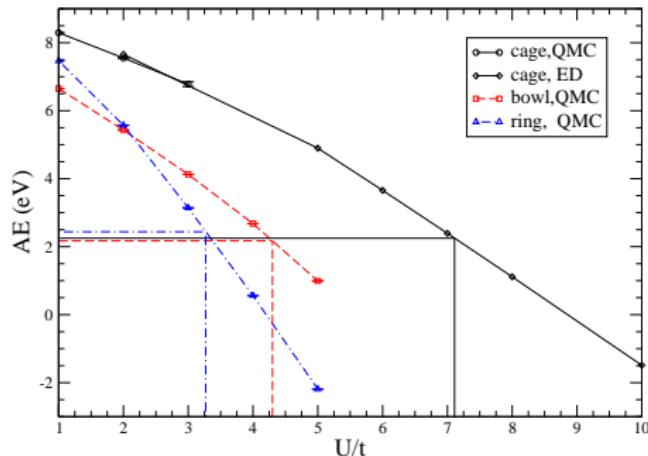
Chow and Friedman [Phys. Rev. B, 2008]

Dynamic structure factor: $S(k, \omega)$.

Effective Hamiltonian Approach

Parameters for C_{20} molecule

- By matching energy gaps from DFT and Huckel energy levels, we can determine the hopping $t \sim 1.36$ eV.
- By matching $\Delta E = E(20) - E(21)$ to experimental value of 2.25 eV, we find $U/t \sim 7.1$. Consistent with U values assumed in some literatures.



Parameters for C_{60} molecule

$t \sim 2.5$ eV

$U/t \sim 4$

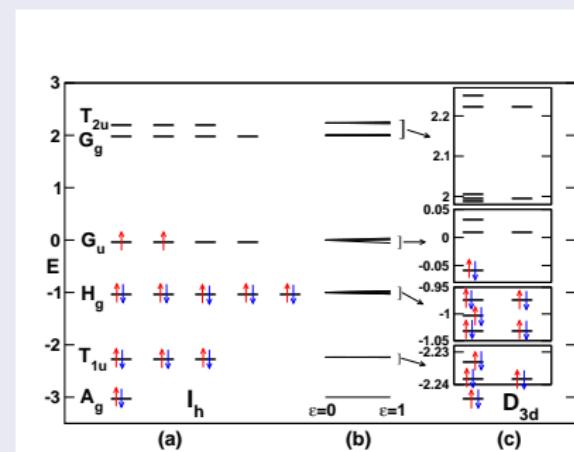
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Single C₂₀ molecule

Magnetic properties

- Magnetic from Huckel energy diagram;
- With ED and QMC:
 - Spin triplet (magnetic) for $U/t < U_c/t \sim 4.10$ (3-fold orbital degeneracy);
 - Spin singlet for $U > U_c$ (non-degenerate); Hund's rule violation;
 - Jahn-Teller is suppressed for $U > U_c$.
- Above for I_h symmetry; similar for D_{3d} symmetry (except at $U = 0$ and $U_c \sim 4.19$).

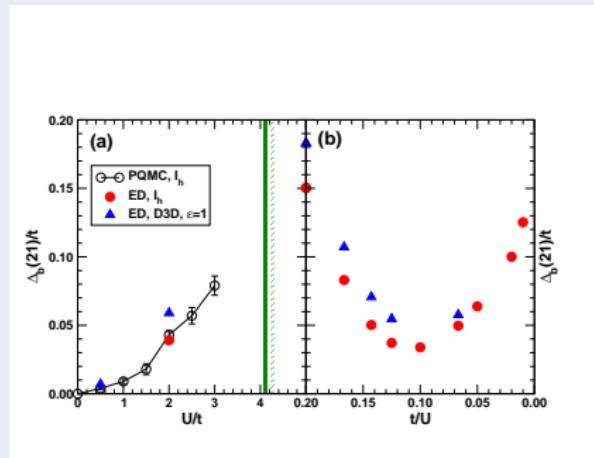


Single C₂₀ molecule

Pair-binding energy

$$\Delta_b(N+1) = E(N+2) - 2E(N+1) + E(N). \quad (3)$$

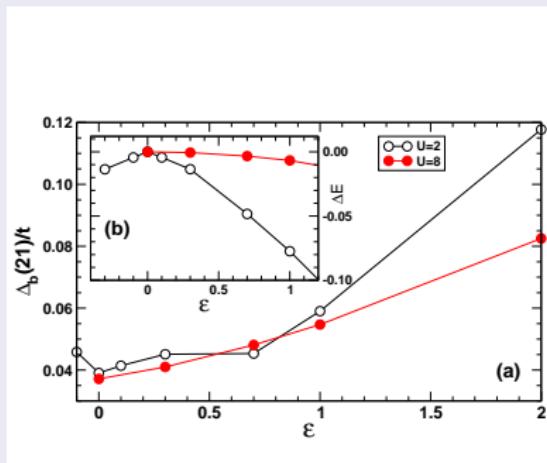
- Importance of mesoscale structure in electronic mechanism of superconductivity, Chakravarty and Kivelson [Science, 1991; Phys. Rev. B, 2001];
- Δ_b always positive \rightarrow absence of pair binding.



Single C₂₀ molecule

Jahn-Teller distortion of C₂₀ with I_h symmetry

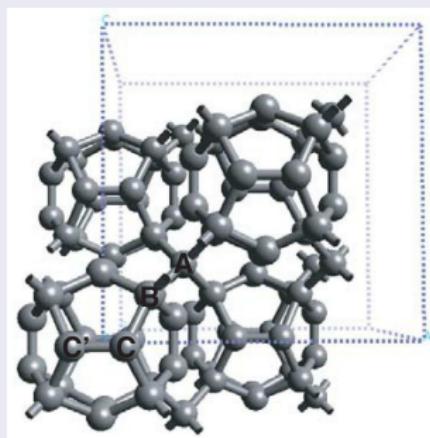
- D_{3d} has 4 bond lengths:
 $a_{ab} = 1.464\text{\AA}$, $a_{bc} = 1.469\text{\AA}$,
 $a_{cc'} = 1.519\text{\AA}$, and
 $a_{cc''} = 1.435\text{\AA}$, Yamamoto et al
[Phys. Rev. Lett., 2005];
- $t_a/t = 1 - \varepsilon(a - \bar{a})/\bar{a}$, where I_h
($\varepsilon = 0$) and D_{3d} ($\varepsilon = 1$);
- Jahn-Teller suppressed for
 $U > U_c$.



C_{20} solid

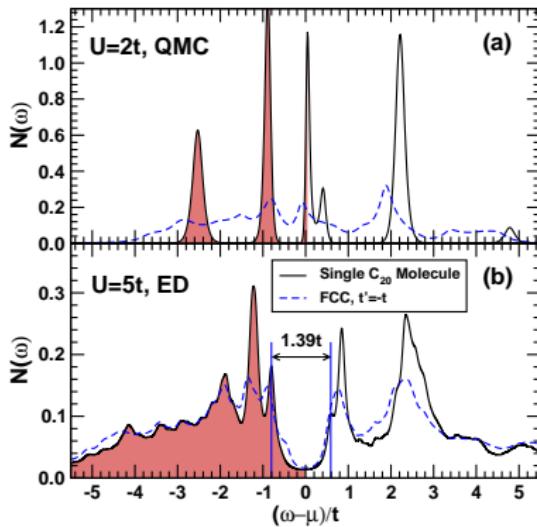
Lattice structure

- Solid C_{20} synthesized by Wang et al [Phys. Lett. A, 2001] and Iqbal et al [Eur. Phys. J. B, 2003].
- fcc structure; 1 additional carbon between 4 C_{20} molecules.
- t intra-molecular hopping; t' inter-molecular hopping.
- Cluster perturbation theory, with exact diagonalization (ED) by Sénéchal et al [Phys. Rev. Lett., 2000]; with QMC by Lin et al [HPC06, p.27].



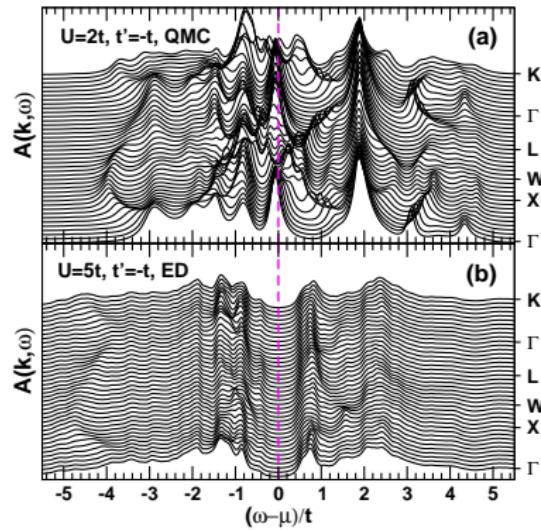
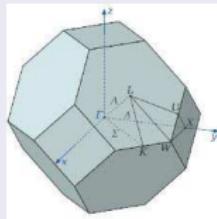
Metal-insulator transition

- Magnetic to nonmagnetic transition at $U = U_c$;
- Metal to insulator transition at $U = U_c$, too;
- Appearance of gap in DOS.



Metal-insulator transition

- Spectral dispersions:



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Single C₆₀ molecule

Pair-binding energy

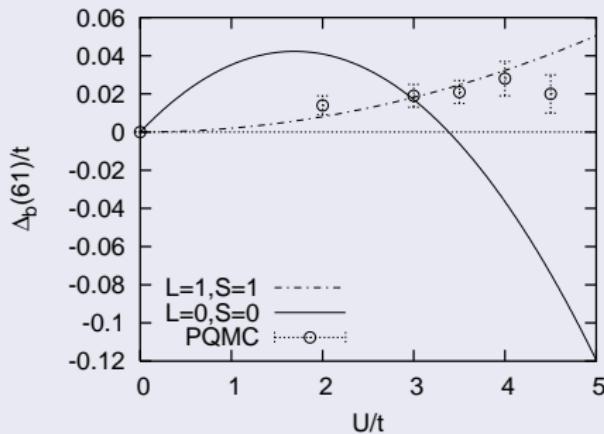
- **Perturbation calculations**

Chakravarty et al [Science 1992];

Ground state of C₆₀²⁻ is in spin-singlet channel at $U/t > 3$;
 $\Delta_b < 0$ for $U/t > 3.3$.

- **QMC calculations**

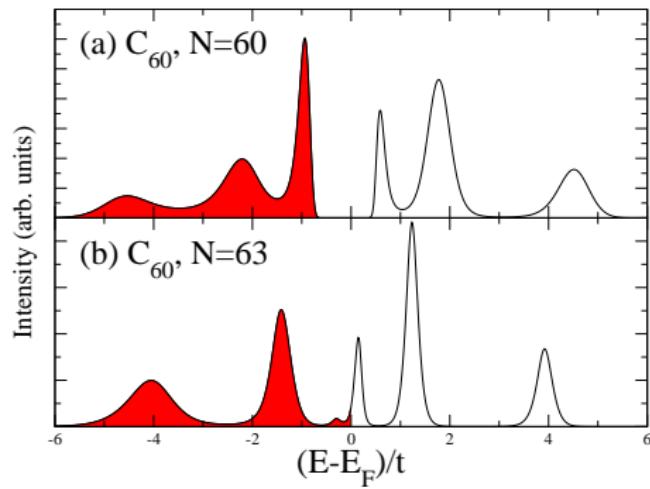
Ground state of C₆₀²⁻ is in spin triplet channel;
 $\Delta_b > 0$ for $U/t \leq 4.5$



Single C₆₀ molecule

Density of states (DOS)

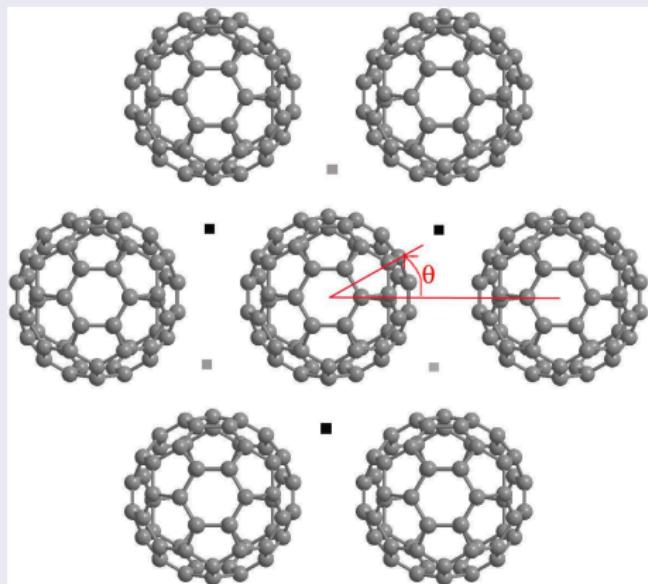
- Neutral C₆₀ molecule is insulating;
- Three electron doped C₆₀ molecule is metallic.



C_{60} monolayer

Monolayer setup

- 2D hexagonal C_{60} superlattice: Z.X. Shen's group [Science, 2003; Phys. Rev. B, 2005]
- Orientation dependent band dispersion.



C₆₀ monolayer

Intermolecular hopping matrix

$$V = - \sum_{\langle Ii, Jj \rangle \sigma} (t' \alpha_{Ii, Jj} + t_{Ii, Jj}^{\text{ind}}) (c_{Ii\sigma}^\dagger c_{Jj\sigma} + h.c.), \quad (4)$$

where **direct hoppings** between NN molecules are given by

$$\alpha_{Ii, Jj} = [V_\sigma(d) - V_\pi(d)](\hat{\mathbf{R}}_{Ii} \cdot \hat{\mathbf{d}})(\hat{\mathbf{R}}_{Jj} \cdot \hat{\mathbf{d}}) + V_\pi(d)(\hat{\mathbf{R}}_{Ii} \cdot \hat{\mathbf{R}}_{Jj}), \quad (5)$$

Satpathy et al [Phys. Rev. B, 1992], and **indirect hoppings** via K⁺ ions are given by

$$t_{Ii, Jj}^{\text{ind}} = \sum_{\gamma} \frac{t_{Ii, \gamma} t_{Jj, \gamma}}{\epsilon_C - \epsilon_K}, \quad (6)$$

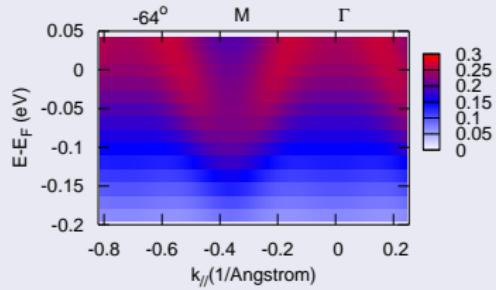
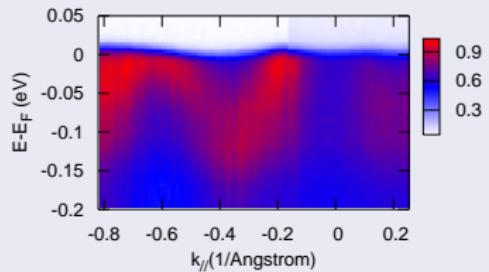
$$t_{Ii, \gamma} = 1.84D \frac{\hbar^2}{md^2} e^{-3(d-R_{\min})}. \quad (7)$$

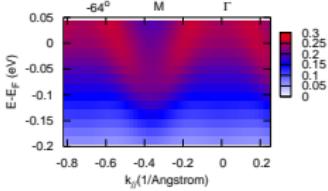
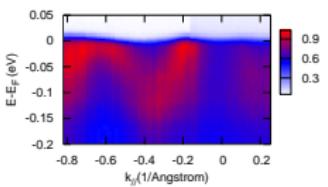
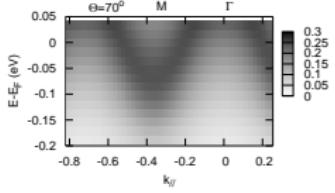
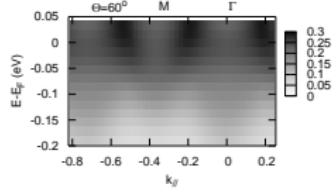
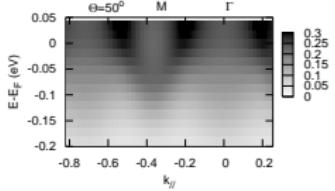
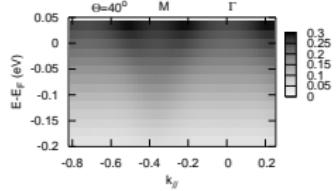
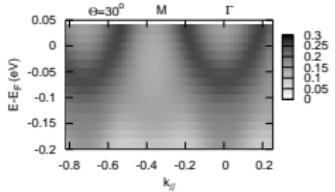
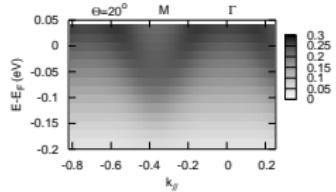
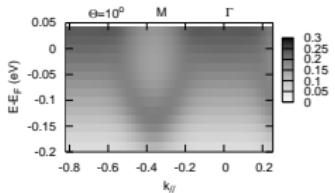
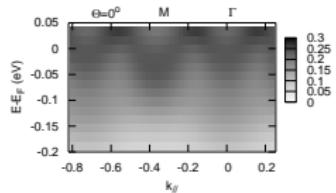
Gunnarsson et al [Phys. Rev. B, 1998].

C_{60} monolayer

Band dispersion matching

- Intermolecular hopping t' and molecular orientation angle θ are variational parameters.
- Cluster perturbation theory.
- Best fit gives $t' = -0.3t$, $\theta = 64^\circ$.

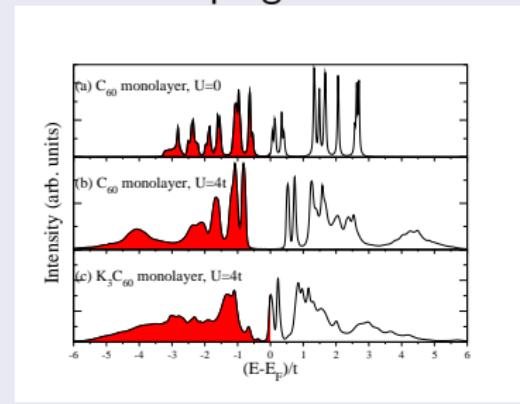




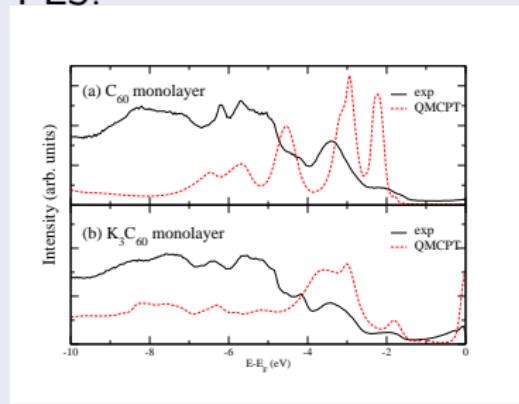
C_{60} monolayer

Monolayer DOS

- Evolution of DOS with U and electron doping.



- Comparison with experimental PES.



Recent developments in experiments

Future projects

- K_4C_{60} monolayer, Crommie group [Science, 2005];
 K_xC_{60} monolayers $3 \leq x \leq 5$, Crommie group [Phys. Rev. Lett., 2007];
 - Why is K_4C_{60} insulating? Jahn-Teller?
 - Role of orientational ordering?
- Cs_3C_{60} superconductor, Rosseinsky group [Science, 2009]
 - $T_c = 38K$; disorder-free
 - parent state: AF insulator; pressure-induced SC
 - unconventional superconductor
- Single molecule transistors: Winkelmann et al [Nature Physics 2009], Yu et al [Nano Lett. 2004], etc.
 - Can we calculate transport property?
- Carbon is light. Intramolecular vibration frequency up to 0.2 eV;
 $E_F = 0.3$ eV.
 - Can we include Carbon dynamics in QMC?

Summary

- Effective tight-binding Hubbard model can capture most of the physics of Fullerene molecules;
- U/t increases with molecular curvature: C_{20} is the most correlated molecule;
- C_{20} molecule/solid undergoes magnetic metal to nonmagnetic insulator transition at $U_c/t \sim 4.1$;
- Neutral C_{20} is an insulator;
- Jahn-Teller distortion is suppressed in C_{20} molecule since $U/t \sim 7.1$;
- Neutral C_{60} is an insulator; K_3C_{60} a metal;
- Not able to find favorable pair binding from Δ_b calculations.
- Able to compare with ARPES for molecular orientation dependent band dispersion.